

Frequency-dependent specific heat of supercooled liquids and glasses

Jürgen Horbach

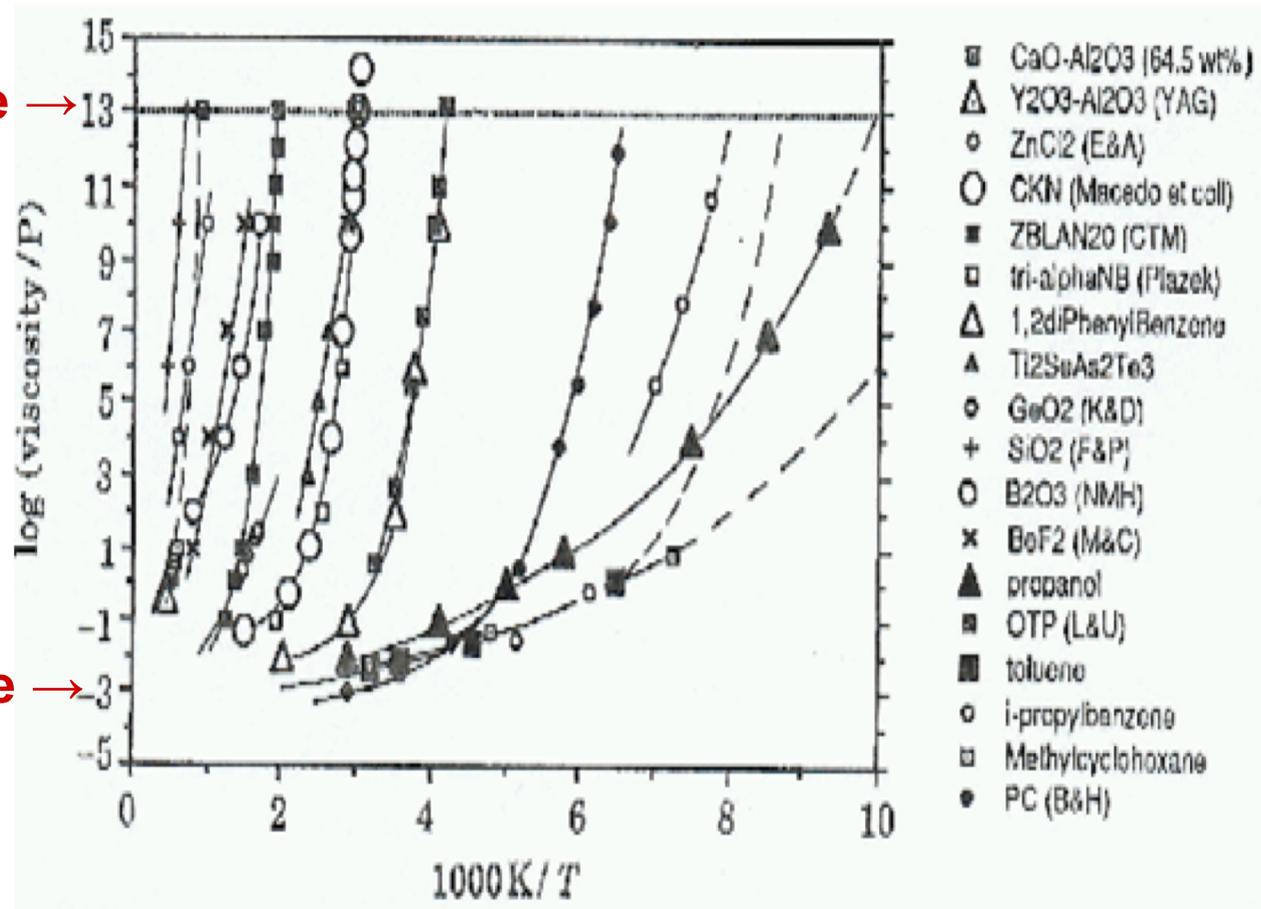
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School Thermodynamics of Glasses

glassforming liquids: shear viscosity

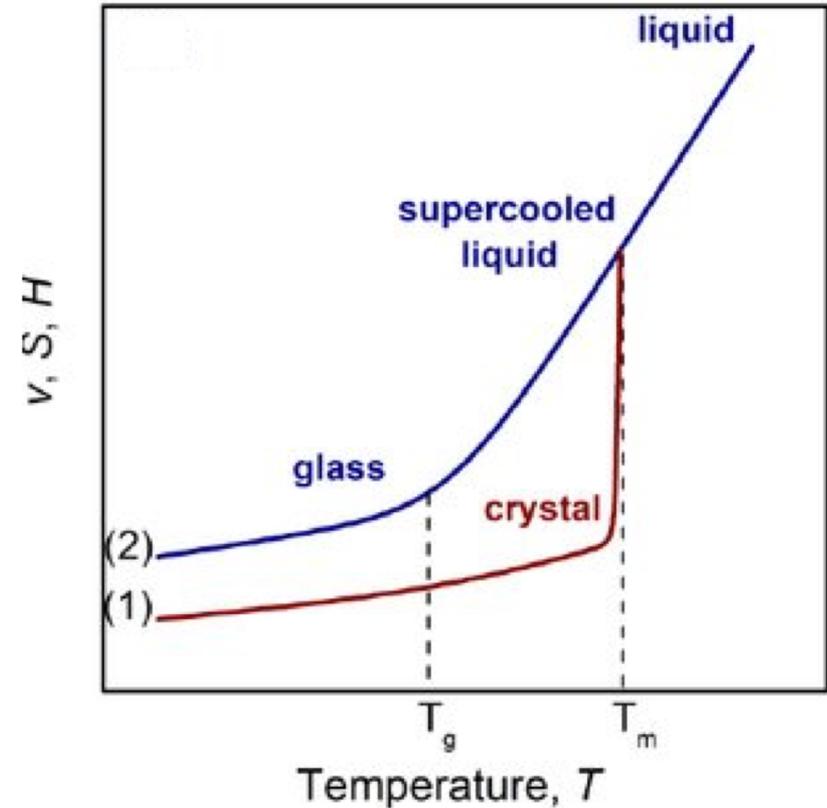
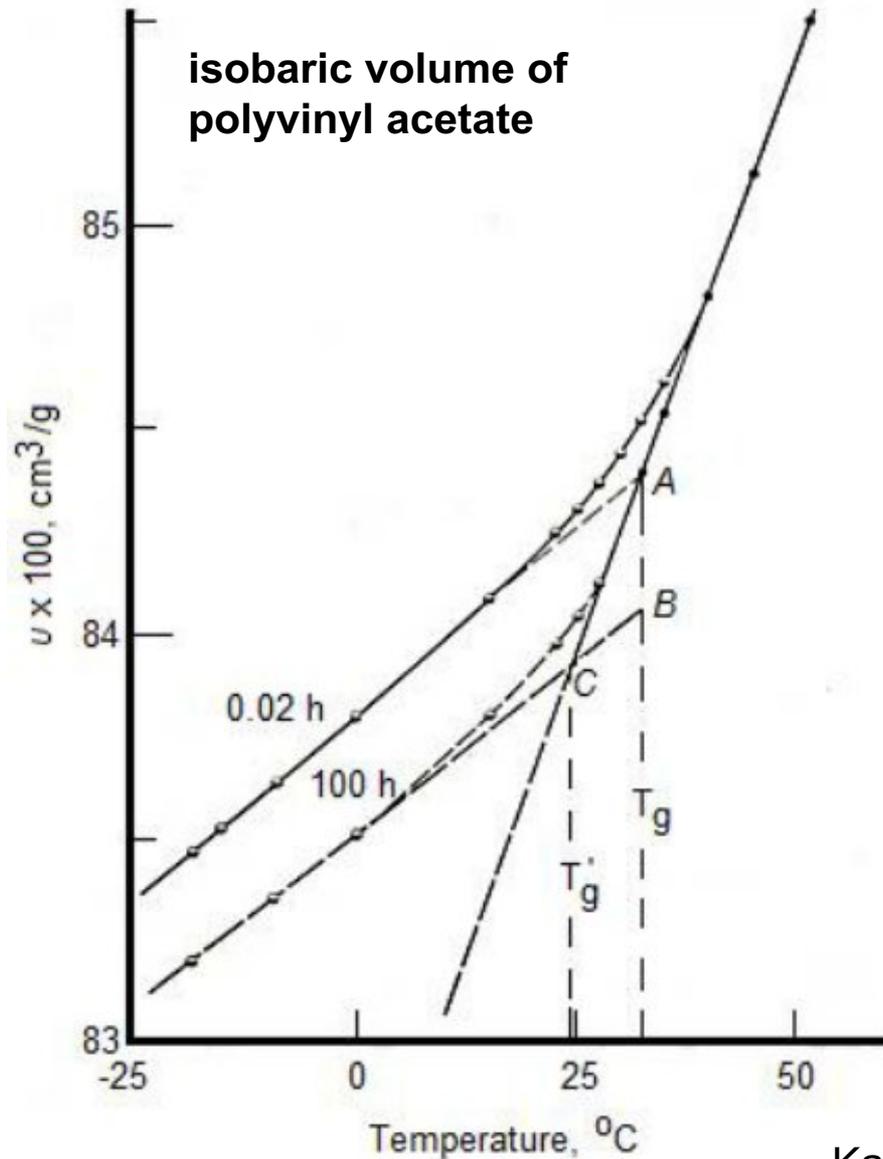
10^{13} Poise →



10^{-3} Poise →

- drastic increase change of dynamics, only small changes in structure
- relation between structural relaxation and thermodynamics?

glass transition: thermodynamic quantities



behavior of 2nd derivatives of thermodynamic potentials?

specific heat

- specific heat at constant volume in canonical ensemble:

$$C_V = \left(\frac{\partial E}{\partial T} \right)_V = \frac{\partial \langle \mathcal{H} \rangle_{\text{can}}}{\partial T} = \frac{1}{k_B T^2} \left(\langle \mathcal{H}^2 \rangle_{\text{can}} - \langle \mathcal{H} \rangle_{\text{can}}^2 \right)$$

- H : Hamilton function of the system
- canonical average:

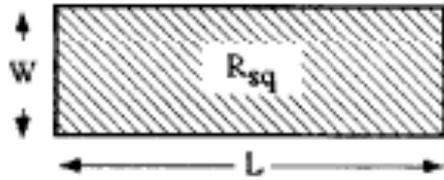
$$\langle \dots \rangle_{\text{can}} = \frac{1}{Z_{\text{can}}} \int \dots e^{-\beta \mathcal{H}} d\Gamma \quad \text{with} \quad Z_{\text{can}} = \int e^{-\beta \mathcal{H}} d\Gamma \quad \text{and} \quad \beta = \frac{1}{k_B T}$$

- Z_{can} : canonical partition function

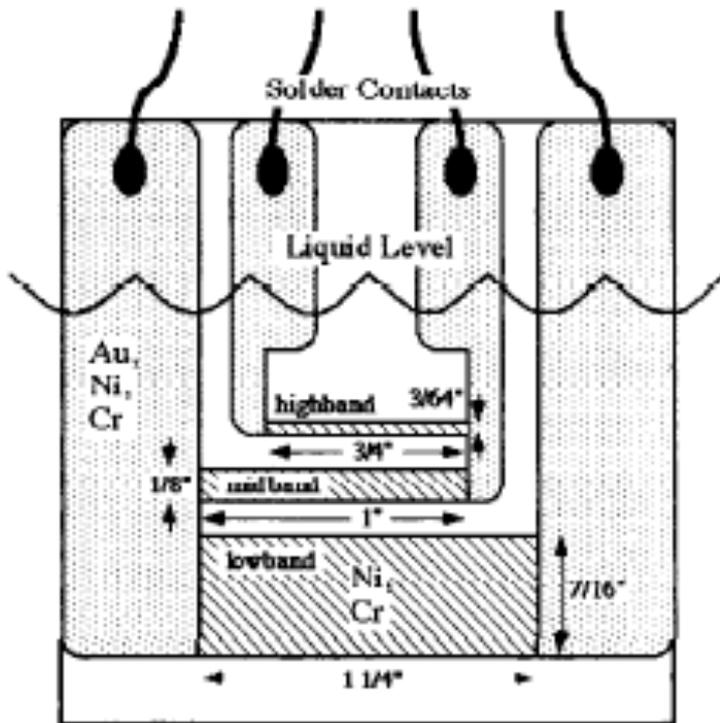
- at constant pressure P in isobaric-isothermal ensemble:

$$C_P = \left(\frac{\partial H}{\partial T} \right)_P \quad \text{with} \quad H: \text{ enthalpy}$$

specific heat spectroscopy



(a)



(b)

- system coupled to a frequency-dependent heat bath
- frequency range of 6 decades
- heat diffusion equation

$$i\omega C_p(\omega)T(x,\omega) = \kappa(\omega) \frac{\partial^2 T(x,\omega)}{\partial x^2}$$

$C_p(\omega)$: specific heat

$\kappa(\omega)$: thermal conductivity

N. O. Birge, S. R. Nagel, *Rev. Sci. Instrum.* **58**, 1464 (1987); N. O. Birge, P. K. Dixon, N. Menon, *Thermochimica Acta* **304/305**, 51 (1997).

frequency-dependent specific heat

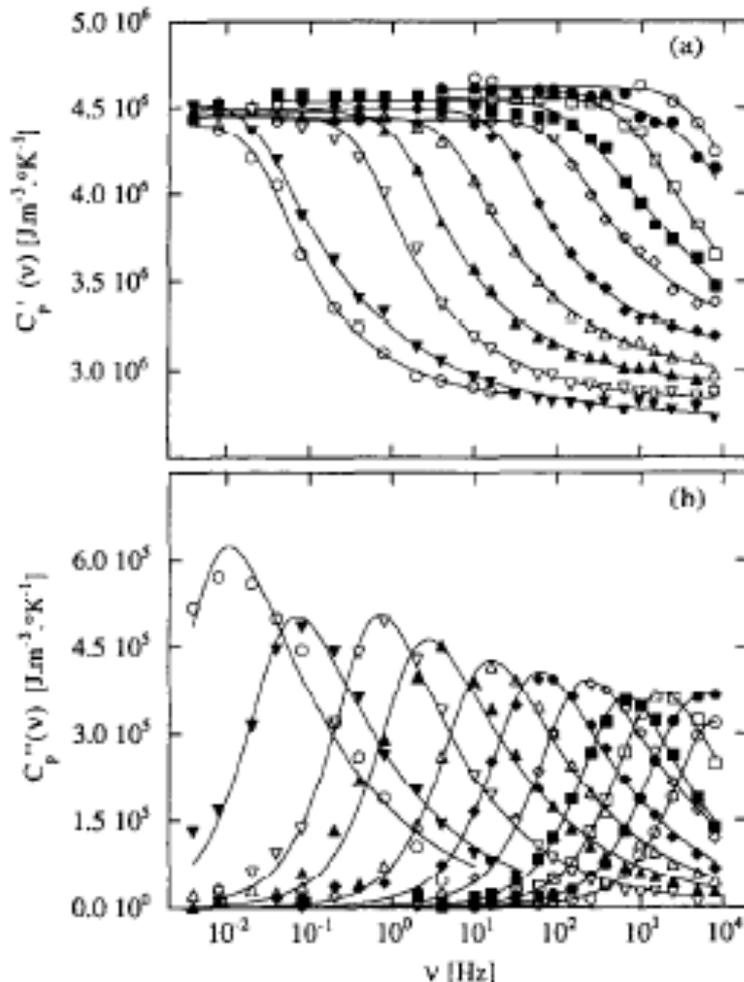


Fig. 6. Real and imaginary parts of $c_p(\nu)$ plotted against $\log_{10}[\nu(\text{Hz})]$ for 11 temperatures from 176.5 to 201 K for the supercooled liquid di-*n*-butylphthalate [12]. The crossover in (a) and the peak in (b) move to lower frequency and become wider at lower T . The lines are fits by a Davidson-Cole form [39].

- real part of frequency-dependent specific heat

$$C_p^I(\nu)$$

- imaginary part of frequency-dependent specific heat

$$C_p^{II}(\nu)$$

- what can one learn from these data?

outline

(1) Molecular Dynamics simulation of silica (SiO_2)

(2) specific heat of glasses

- harmonic approximation, vibrational density of states
- Boson peak

(3) frequency-dependent specific heat in undercooled melts

- real and imaginary part
- thermal diffusivity

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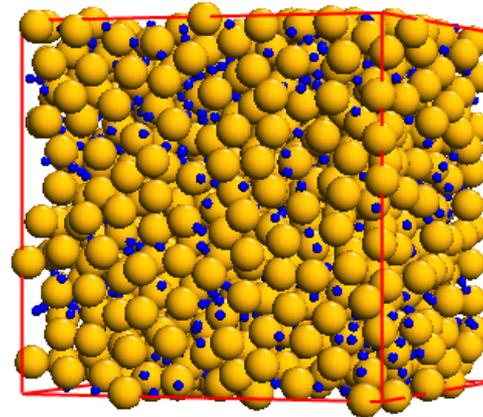
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Molecular Dynamics simulation

- classical system of N particles at positions $\{\vec{r}_i\}$, $i = 1, \dots, N$

$$m_i \ddot{\vec{r}}_i = - \frac{\partial U_{\text{pot}}}{\partial \vec{r}_i} = \vec{F}_i$$



- U_{pot} : potential function
- simplest case: pairwise additive interaction between point particles

$$U_{\text{pot}} = \sum_{i=1}^{N-1} \sum_{j>i}^N u(r_{ij}) \quad r_{ij} = \vec{r}_i - \vec{r}_j$$

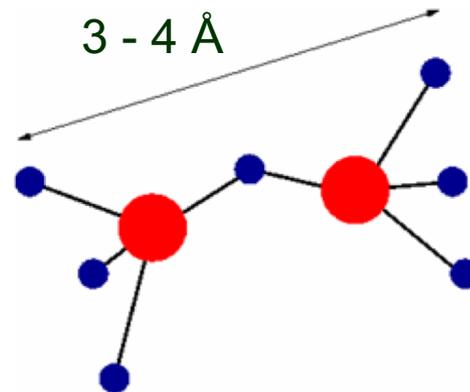
- solution of equations of motion yield trajectories of the particles, i.e. positions and velocities of all the particles as a function of time

simulation details: SiO₂

- BKS potential for silica:
$$u_{\alpha\beta}(r) = \frac{q_{\alpha}q_{\beta}e^2}{4\pi\epsilon r} + A_{\alpha\beta}\exp(-B_{\alpha\beta}r) - \frac{C_{\alpha\beta}}{r^6}$$
$$\alpha\beta = \text{SiSi}, \text{SiO}, \text{OO}$$

- parameters $A_{\alpha\beta}$, $B_{\alpha\beta}$, $C_{\alpha\beta}$ in van Beest, Kramer, van Santen, *Phys. Rev. Lett.* **64**, 1955 (1990).
- $N = 336 - 8064$, simulations at constant density $\rho = 2.36 \text{ g/cm}^3$
- undercooled melt $6100 \text{ K} \geq T \geq 2750 \text{ K}$; glass at 300 K (obtained by runs from 2750 K with a cooling rate of 10^{12} K/s)

- tetrahedral network structure



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specific heat for harmonic solid

- in harmonic approximation:
$$C_V = \frac{h^2}{k_B T^2} \int_0^\infty \frac{\nu^2 \exp(\beta h \nu)}{(\exp(\beta h \nu) - 1)^2} g(\nu) d\nu$$

- classical limit: Dulong-Petit law:
$$C_V = 3N_A k_B$$

- $g(\nu)$: vibrational density of states (VDOS), with ν the frequency

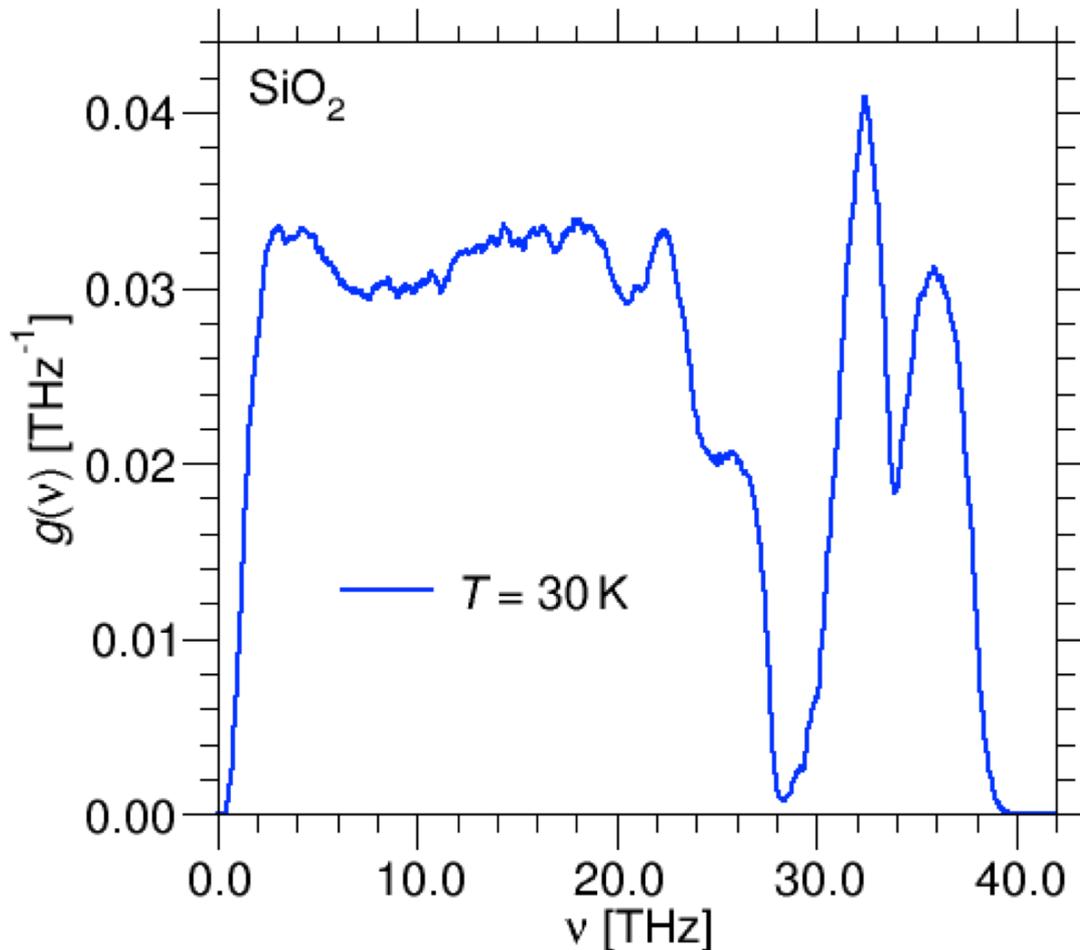
- VDOS can be obtained from velocity autocorrelation function (VACF):

$$g(\nu) = \frac{1}{Nk_B T} \sum_{j=1}^N \int_{-\infty}^{\infty} dt m_j \exp(i2\pi\nu t) \langle \vec{v}_j(t) \cdot \vec{v}_j(0) \rangle$$

- low temperatures: Debye theory

$$g_D(\nu) = \frac{3\nu^2}{\nu_D^3} \quad \text{with} \quad \nu_D = \left(\frac{9N}{4\pi V} \right)^{1/3} \left(\frac{2}{c_T^3} + \frac{1}{c_L^3} \right)^{-1/3} \Rightarrow C_V(T) \propto T^3$$

vibrational density of states (VDOS)



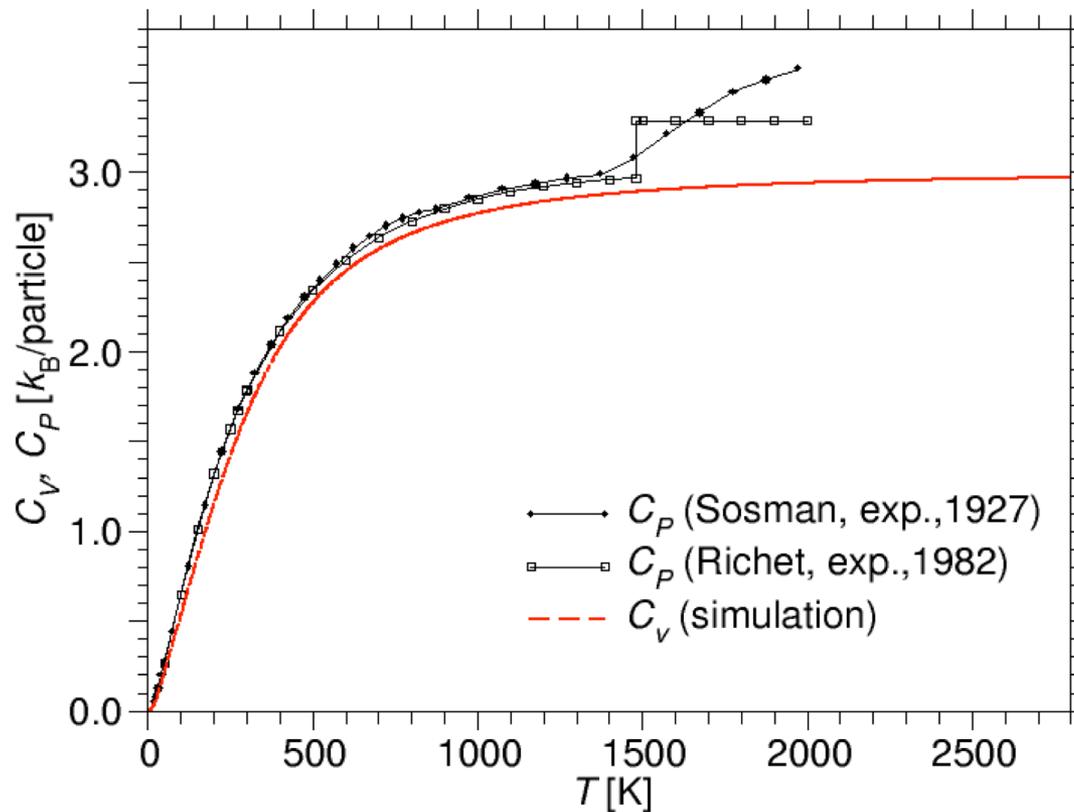
$g(\nu)$ for BKS silica:

- purely classical calculation
- $\nu > 30$ THz: stretching modes in single SiO₄ unit
- intermediate frequency band not in good agreement with ab initio calculations

J. Horbach, W. Kob, K. Binder, *J. Phys. Chem. B* **103**, 4104 (1999); P. Scheidler, W. Kob, A. Latz, J. Horbach, K. Binder, *Phys. Rev. B* **63**, 104204 (2001).

comparison to experiment

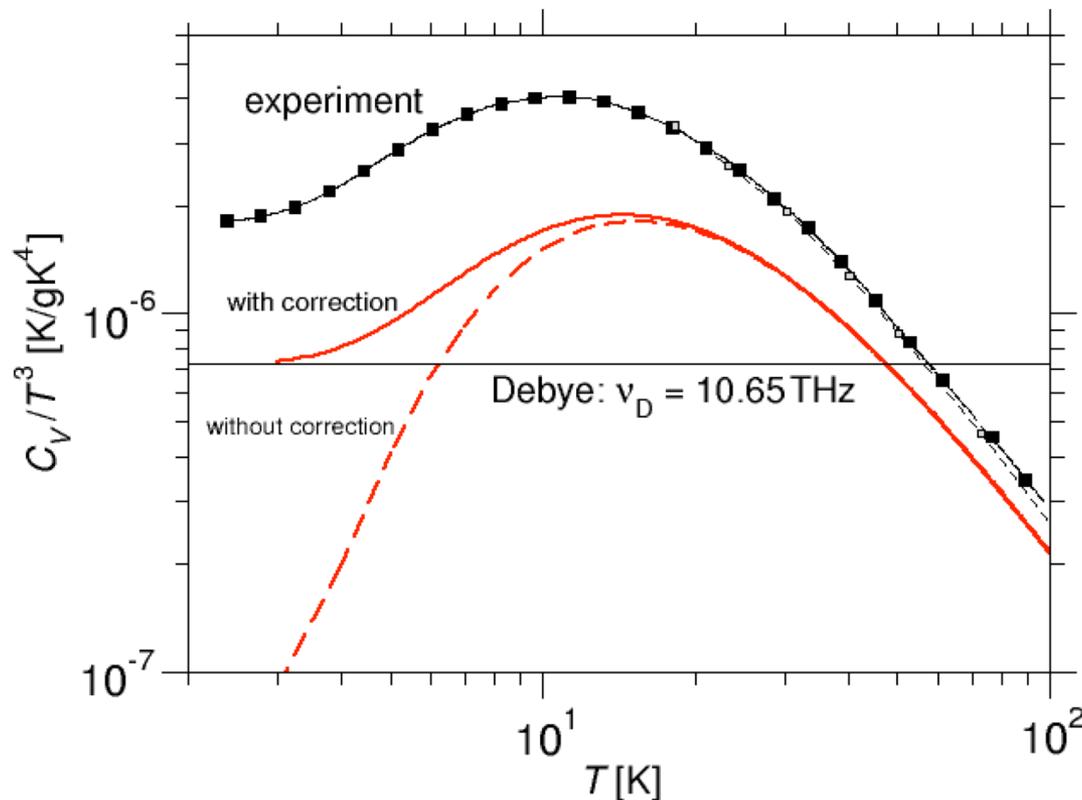
$$C_V = \frac{h^2}{k_B T^2} \int_0^\infty \frac{v^2 \exp(\beta h v)}{(\exp(\beta h v) - 1)^2} g(v) dv$$



- good agreement between simulation and experiment below T_g
- deviations to the experiment for $T > T_g$ mainly due to structural relaxation (and anharmonicities)
- behavior at low temperatures: $T < 50$ K

low temperatures: Boson peak

$$g_D(\nu) = \frac{3\nu^2}{\nu_D^3} \quad \text{with} \quad \nu_D = \left(\frac{9N}{4\pi V} \right)^{1/3} \left(\frac{2}{c_T^3} + \frac{1}{c_L^3} \right)^{-1/3} \quad \Rightarrow \quad C_V(T) \propto T^3$$



- correction: add $g_D(\nu)$ to measured $g(\nu)$ for $\nu \leq 0.73$ THz

$$c_L = 6161 \text{ m/s}$$

$$c_T = 3915 \text{ m/s}$$

$$\Rightarrow \nu_D = 10.65 \text{ THz}$$

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q-dependent temperature fluctuation

- microscopic temperature of a particle:

$$\theta(\vec{p}_i) = \frac{1}{3k_B m} \vec{p}_i^2 \quad \text{with } \vec{p}_i : \text{ momentum of particle } i$$

- microscopic temperature field density:

$$T(\vec{r}, t) = \sum_{i=1}^N \theta(\vec{p}_i) \delta(\vec{r} - \vec{r}_i(t)), \quad \text{FT: } T_q(t) = \sum_{i=1}^N \theta(\vec{p}_i) \exp(i\vec{q} \cdot \vec{r}_i)$$

- temperature fluctuation in Fourier space:

$$\delta T_q(t) = T_q(t) - T \rho_q(t)$$

central quantity: ACF of kinetic energy

- static correlation function independent of q :

$$S_{TT}^c = \langle \delta T_q^*(0) \delta T_q(0) \rangle = \frac{2}{3} N T^2$$

- time-dependent autocorrelation function (ACF):

$$\Phi_{TT}(q, t) = \langle \delta T_q^*(t) \delta T_q(0) \rangle$$

$$q \rightarrow 0 : \Phi_{TT}(t) = \langle \delta T_0^*(t) \delta T_0(0) \rangle$$

- normalized ACF of kinetic energy:

$$K(t) = \frac{\Phi_{TT}(t)}{S_{TT}^c}$$

→ $K(t)$ related in harmonic approximation to VACF

→ Laplace transform of $K(t)$ related to frequency-dependent specific heat

→ $\Phi_{TT}(q, t)$ related to thermal conductivity

relation between ACF of kinetic energy vs. VACF

- VACF weighted with mass m : $J(t) = m \langle \vec{v}(t) \cdot \vec{v}(0) \rangle$
- in harmonic approximation the following relation between ACF of the kinetic energy, $K(t)$, and $J(t)$ holds:

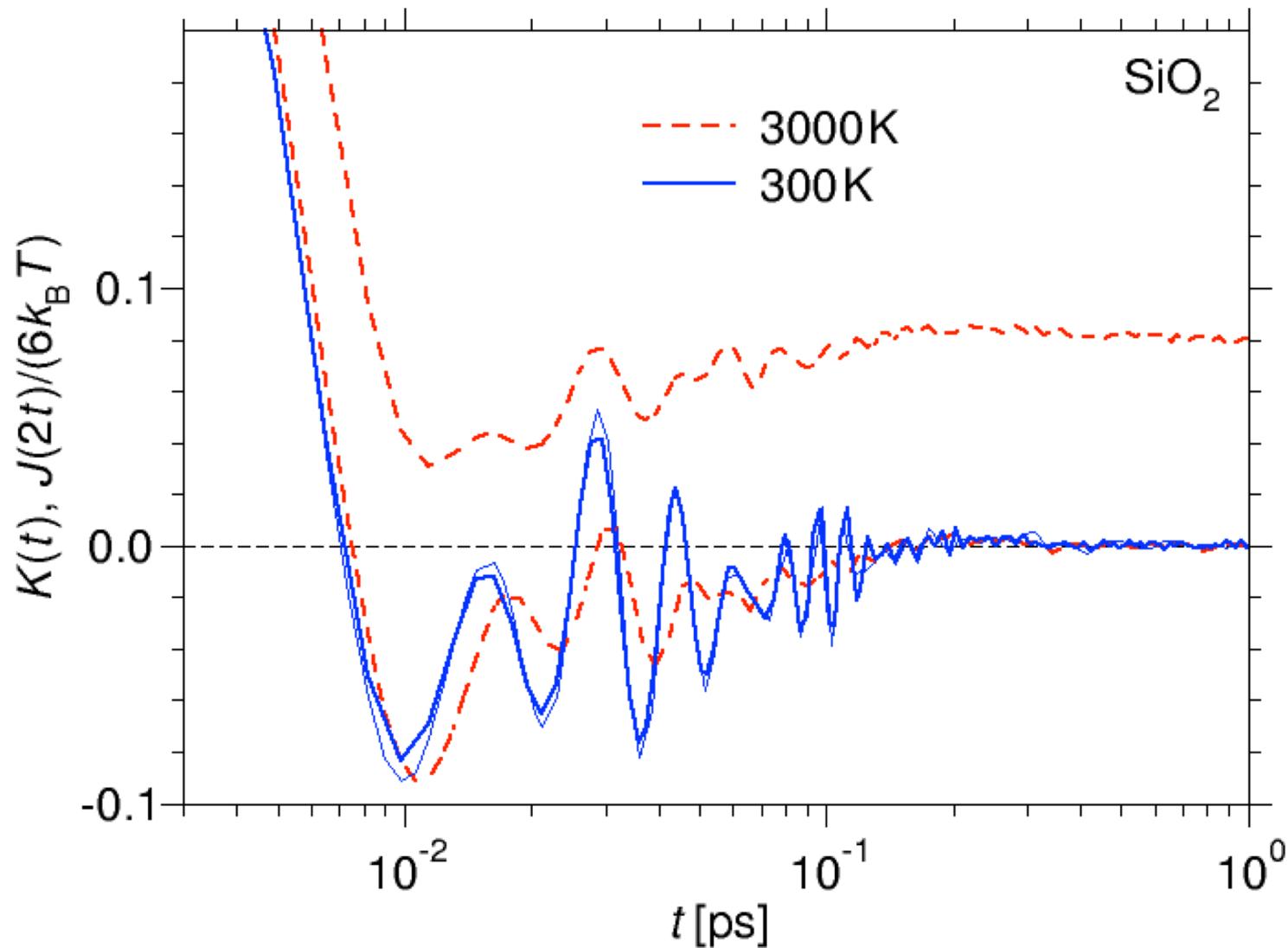
$$K(t) = \frac{J(2t)}{6k_{\text{B}}T}$$

- this implies a relation between $K(\omega)$ and VDOS, $g(\omega)$:

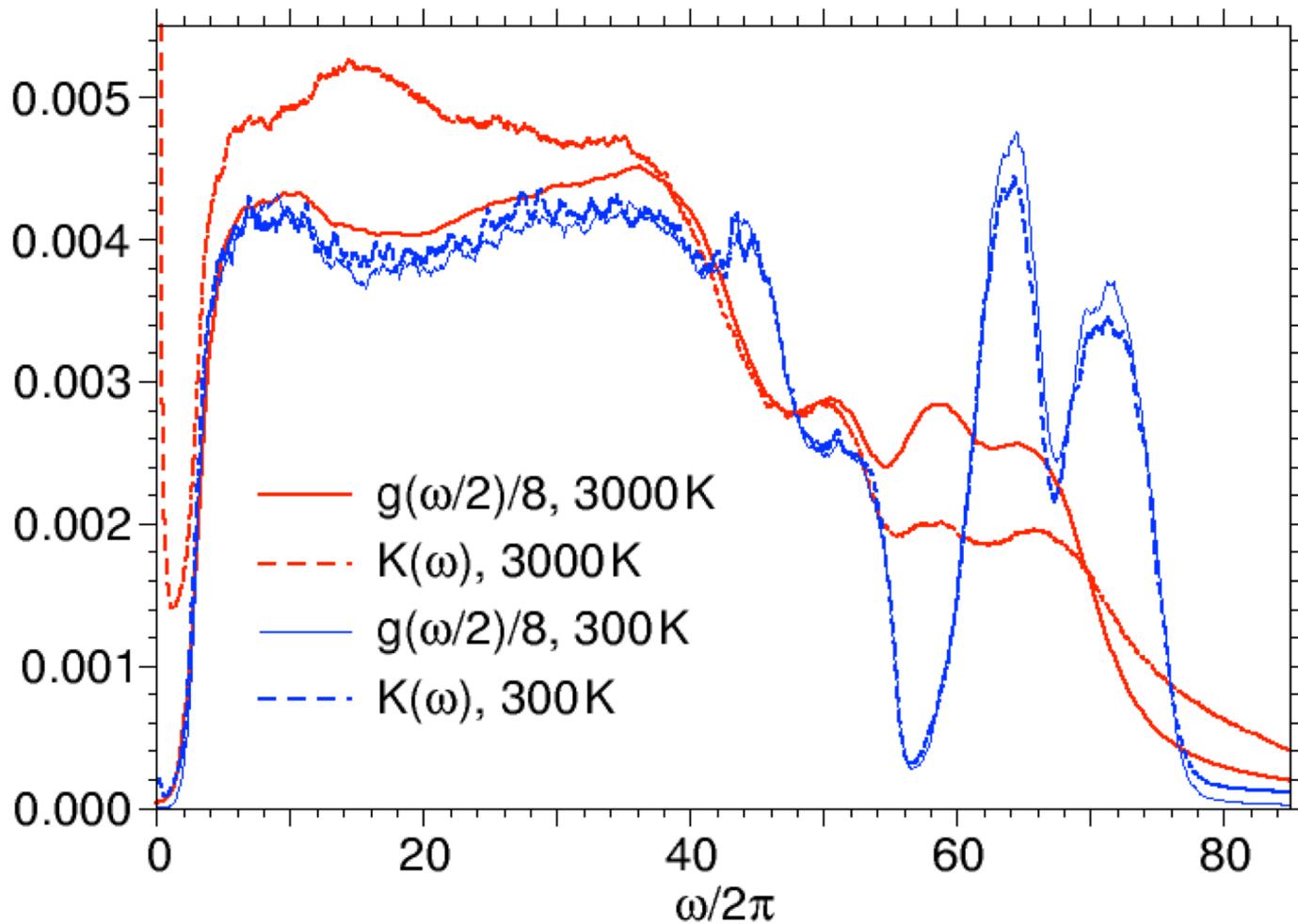
$$K(t) = \frac{J(2t)}{6k_{\text{B}}T} \quad \text{and} \quad \hat{K}(\omega) = \frac{1}{8} g(\omega / 2)$$

- consistent with simulation?

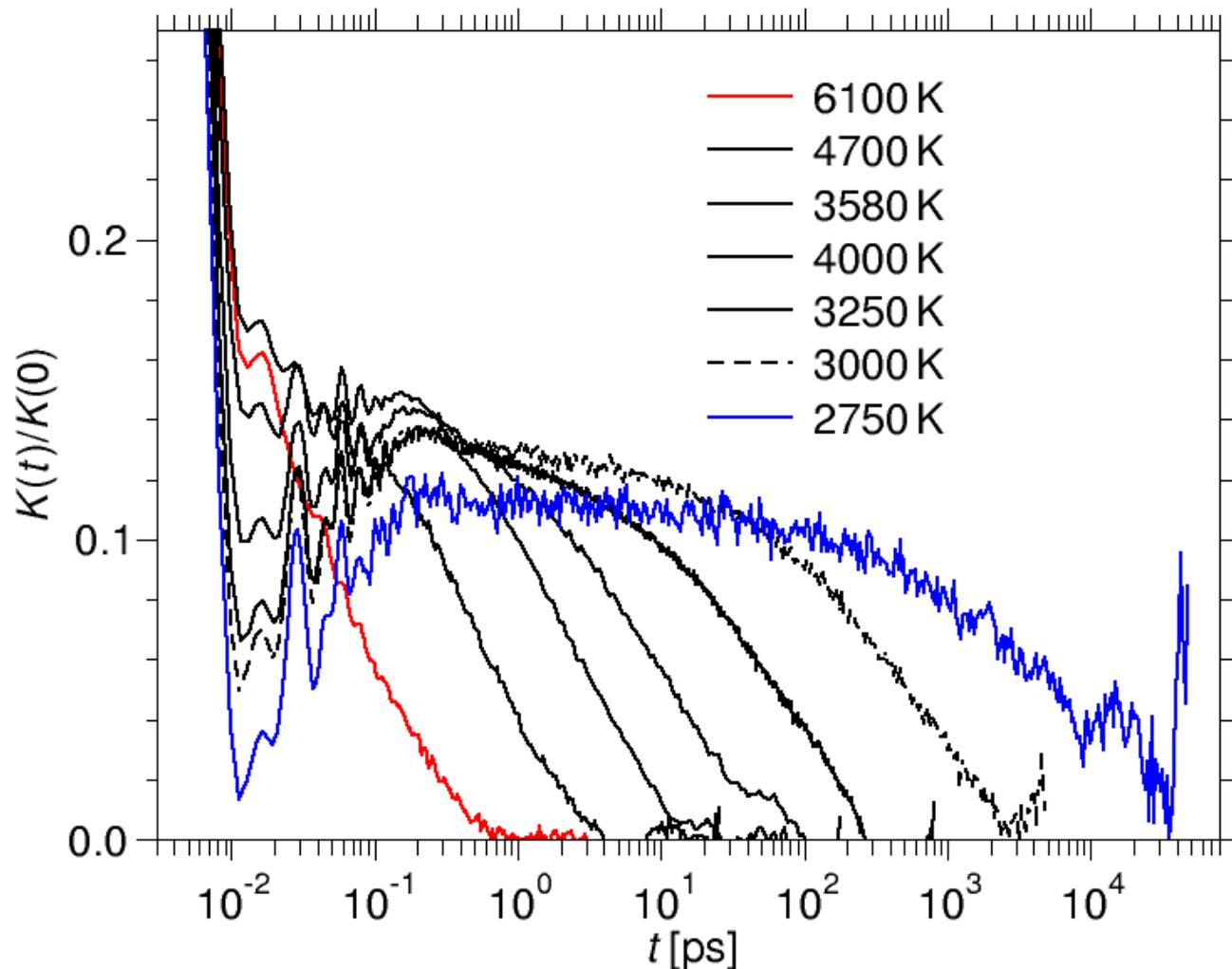
ACF of kinetic energy vs. VACF: harmonic approximation



ACF of kinetic energy vs. VDOS: harmonic approximation?



ACF of kinetic energy at high temperature



→ two-step decay indicates structural relaxation, relation to specific heat?

frequency-dependent specific heat

- static specific heat per particle in canonical ensemble:

$$c_V = \frac{1}{Nk_B T^2} \left(\langle \mathcal{H}^2 \rangle_{\text{can}} - \langle \mathcal{H} \rangle_{\text{can}}^2 \right)$$

- microcanonical ensemble:

$$c_V = \frac{3k_B / 2}{1 - K(0)}$$

- generalization to frequency-dependent specific heat:

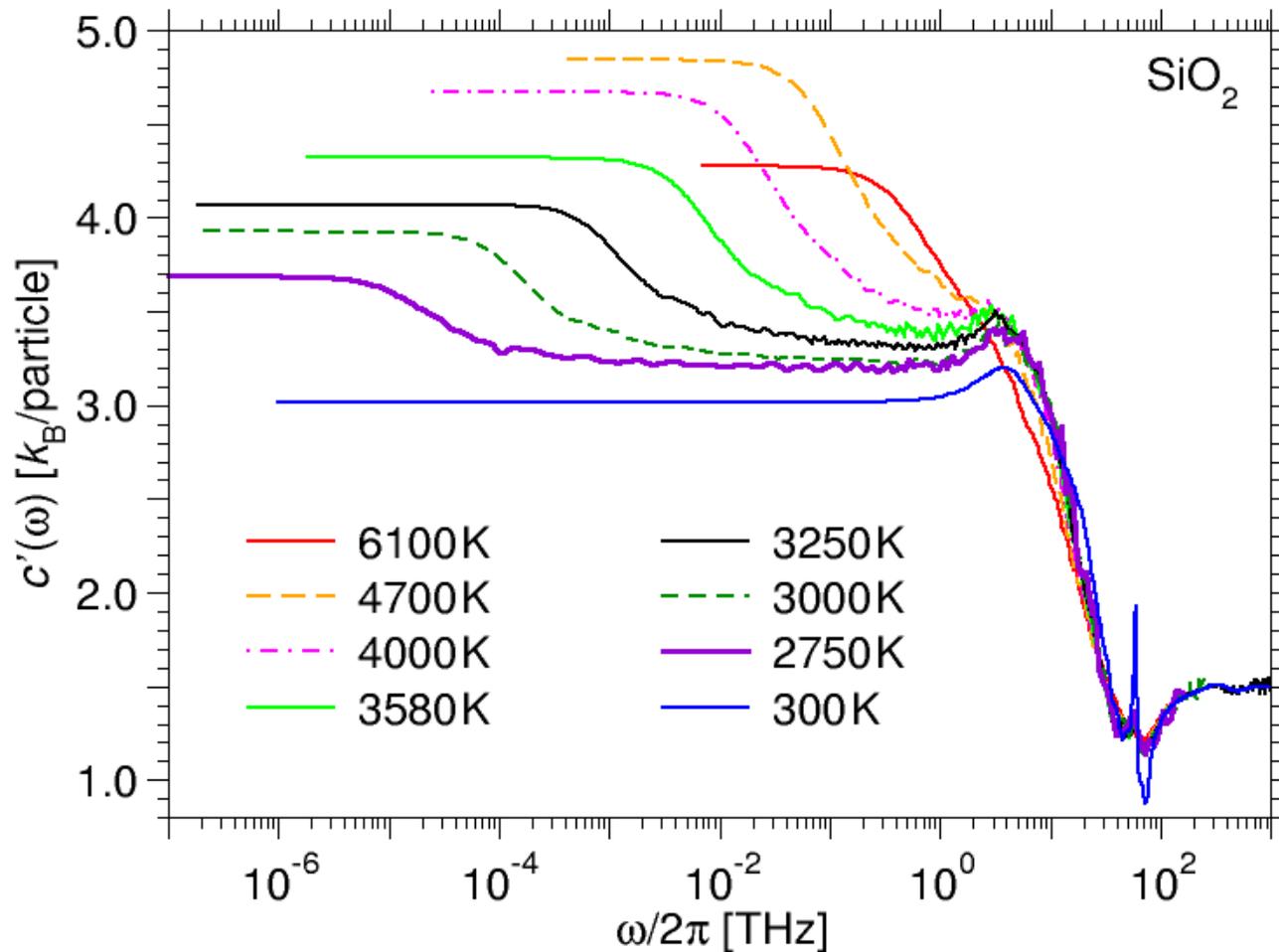
$$c_V(z) = \frac{3k_B / 2}{1 - K(0) - z \text{LT}[K(t)](z)}$$

LT[K(t)](z): Laplace transform of K(t)

- Fourier transform: $z = i\omega$

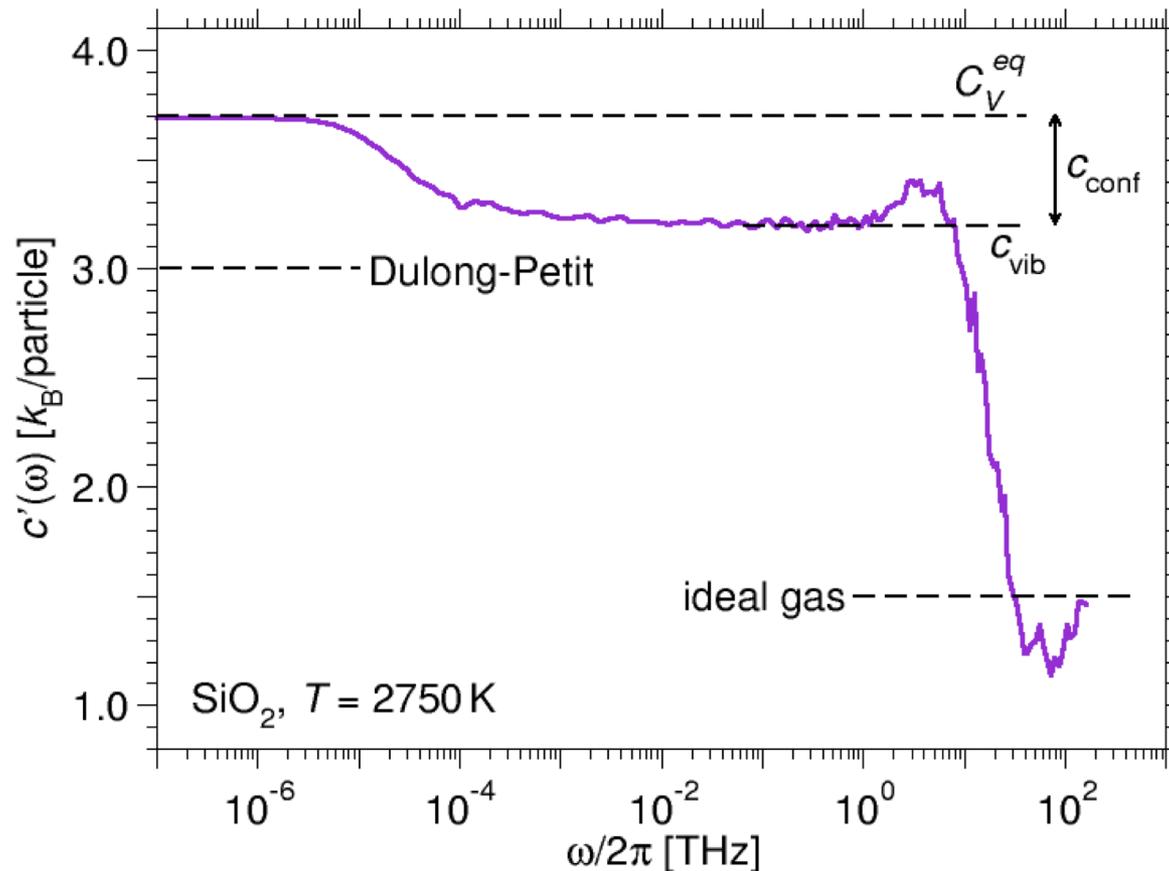
$$c_V(\omega) = c'(\omega) + ic''(\omega)$$

real part of frequency-dependent specific heat



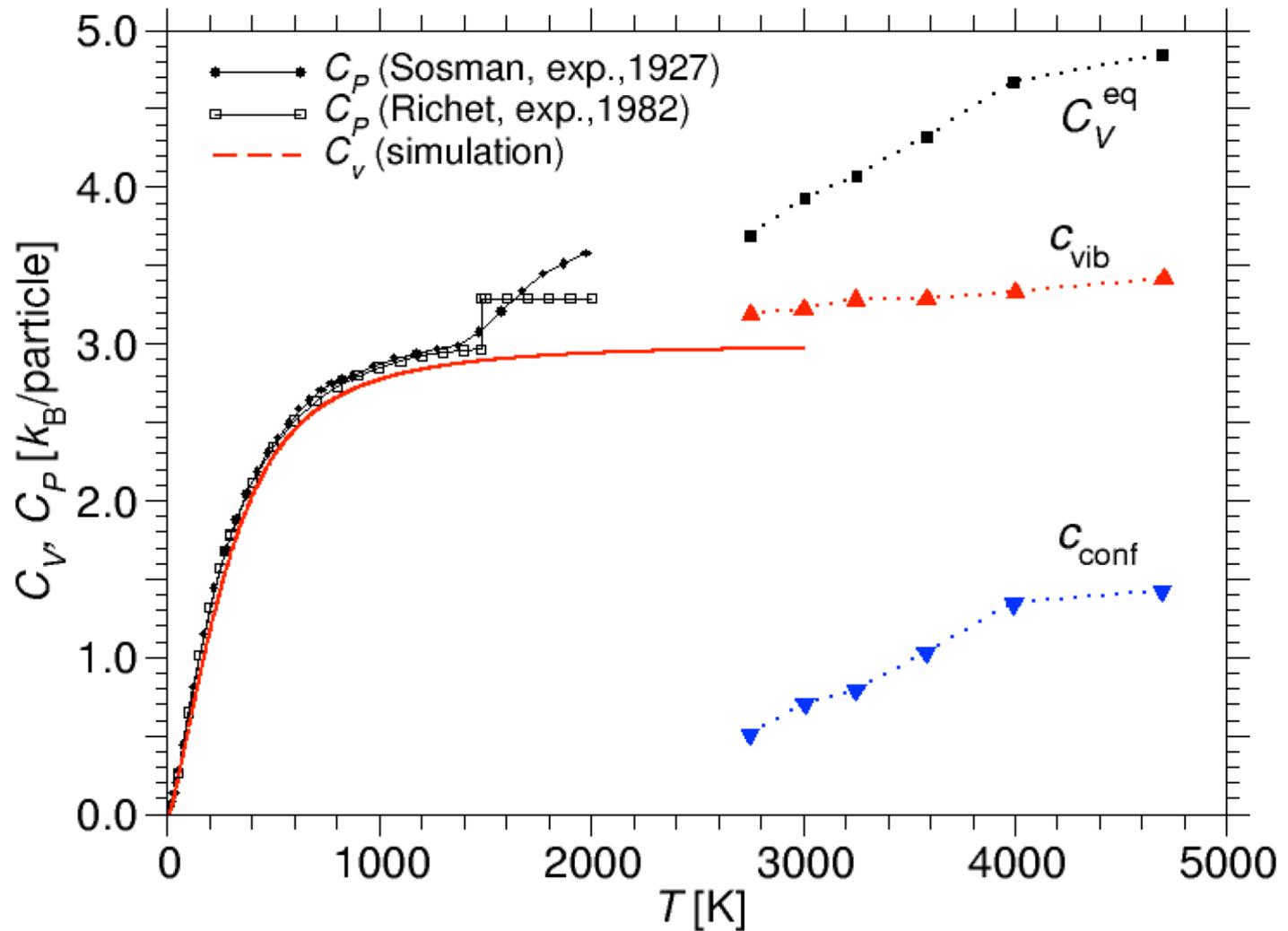
- static specific heat at low ω : with decreasing T this regime shifts to lower frequencies

configurational specific heat

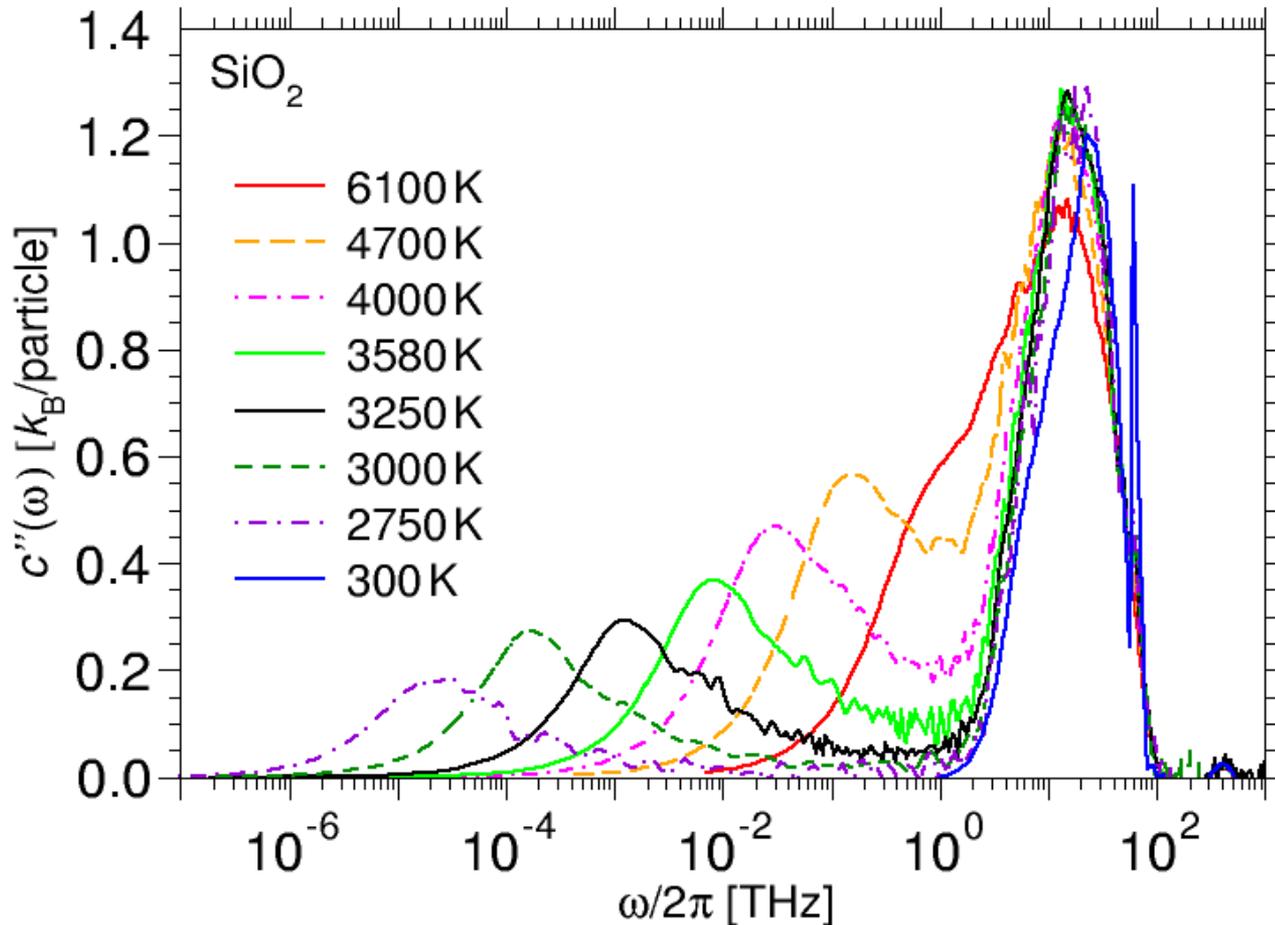


- define “configurational” specific heat c_{conf} :
$$c_{\text{conf}} = C_V^{\text{eq}} - C_{\text{vib}}$$
- vibrational part c_{vib} remains below the glass transition temperature

specific heat: comparison to experiment II

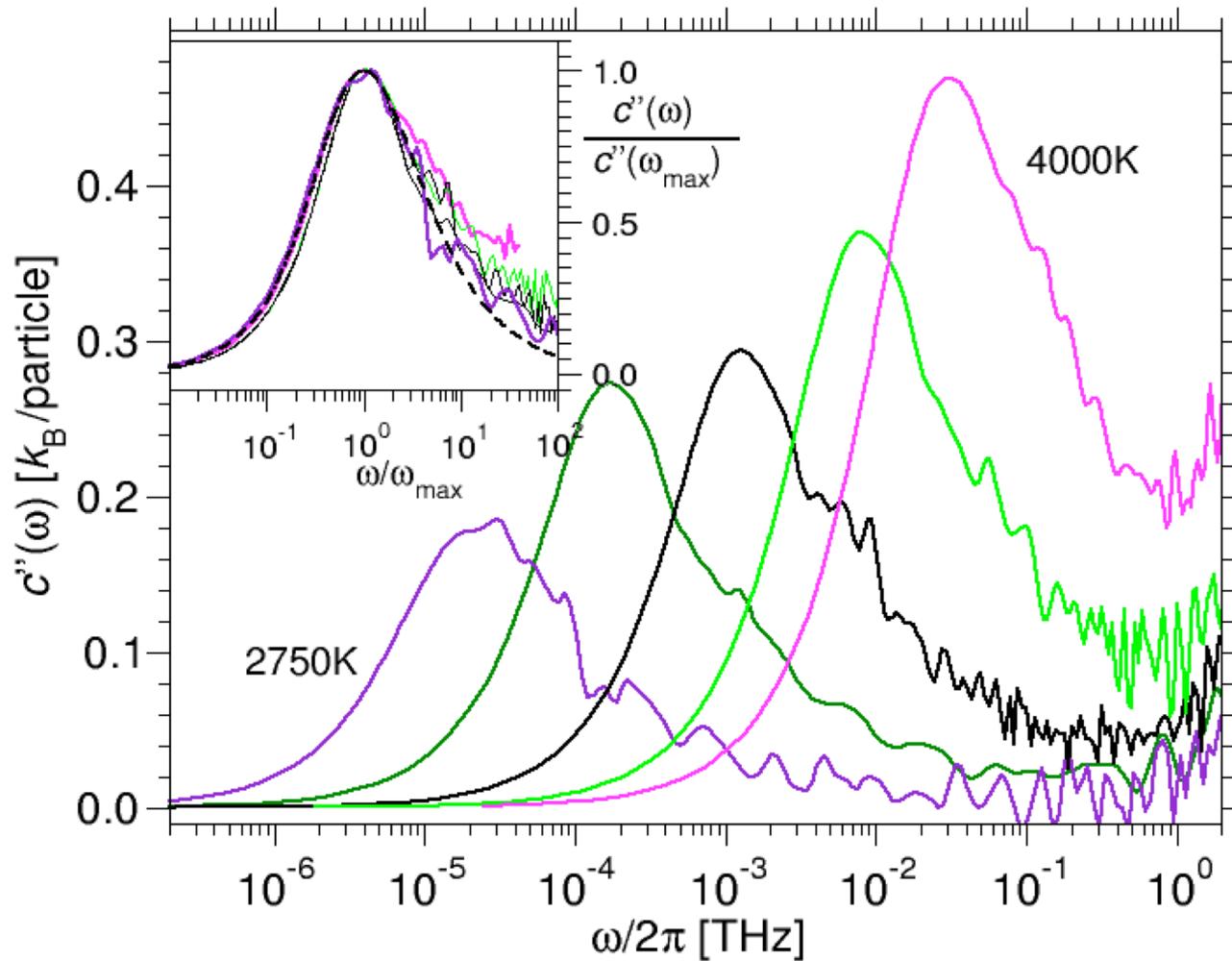


imaginary part of frequency-dependent specific heat



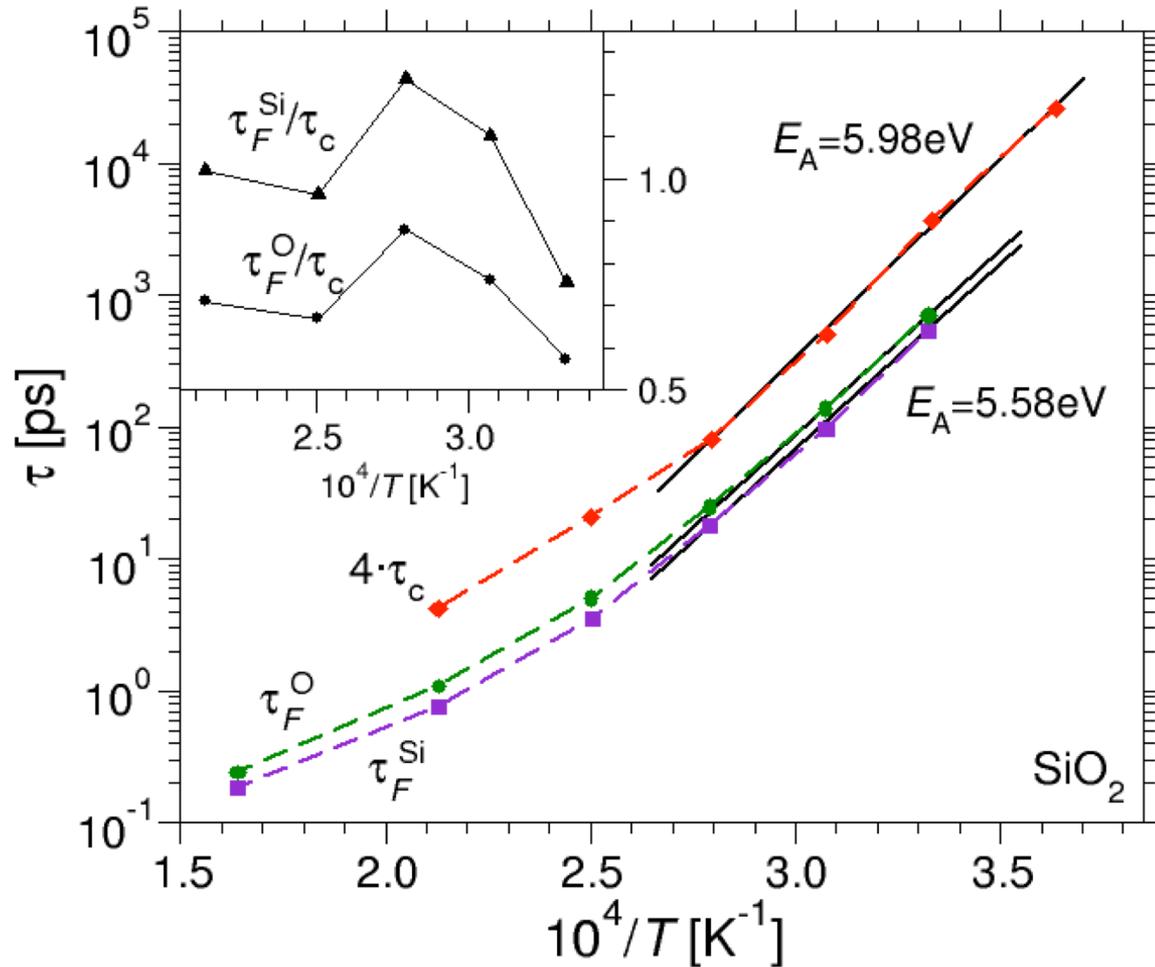
- spectrum with α peak at low frequencies and microscopic peak around 10 THz

the α relaxation



- define α relaxation time: $\tau_c = 1/\omega_{\max}$

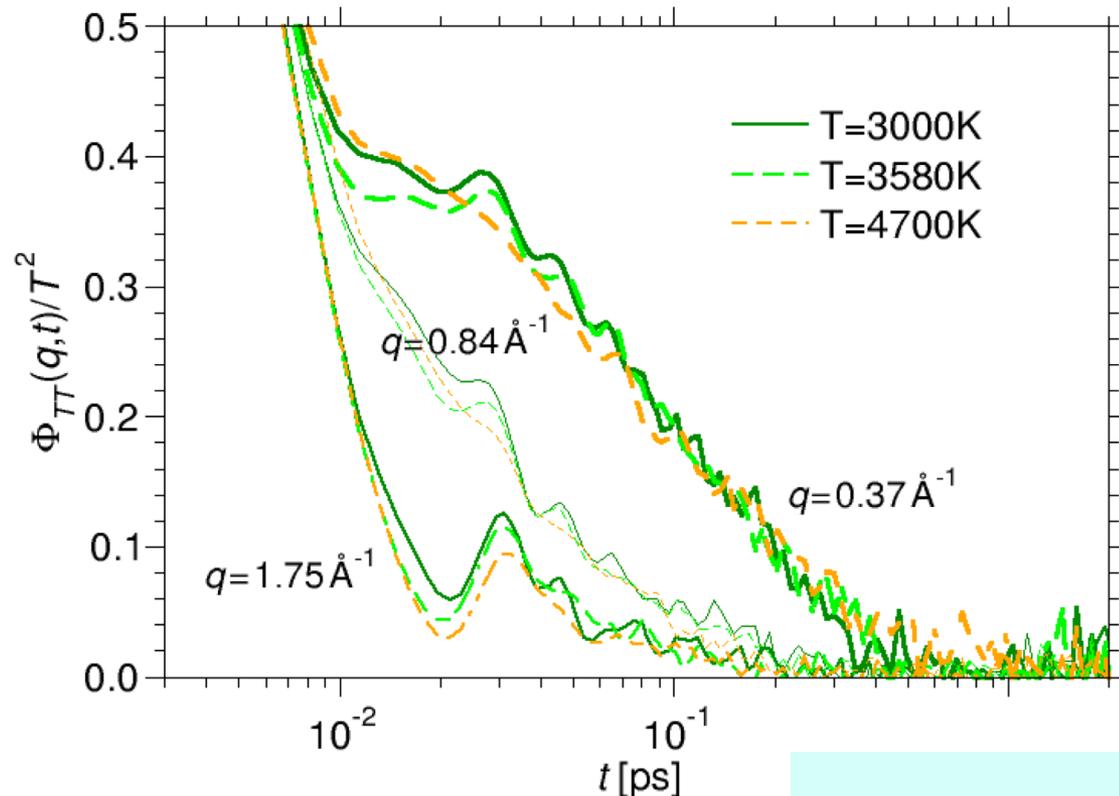
α relaxation times



- relaxation times τ_F obtained from incoherent intermediate scattering functions for Si and O

autocorrelation function of q -dependent temperature

$$\Phi_{TT}(q,t) = \langle \delta T_q^*(t) \delta T_q(0) \rangle$$



T dependence weak:
no coupling to
structural relaxation

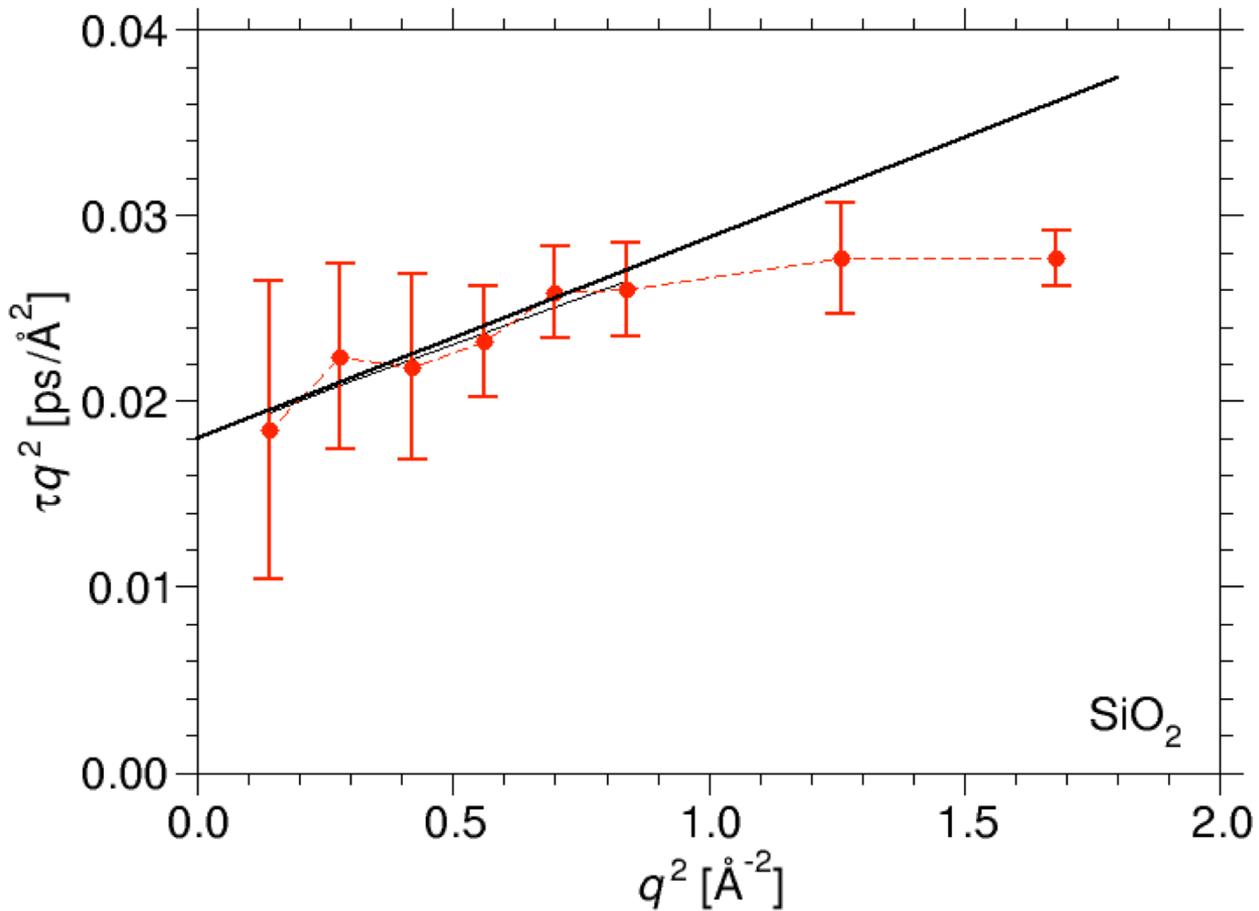
$$\Phi_{TT}(q,t) \propto T^2$$

exponential decay at long
times:

$$\Phi_{TT}(q,t) \propto \exp(-t/\tau_q) \quad \text{with} \quad \tau_q = \frac{1}{D_T q^2} + c$$

$$D_T = \frac{\kappa}{nc_p^{\text{eq}}} \quad \text{thermal diffusivity}$$

thermal conductivity



- thermal conductivity from fit: $\kappa = 2.4$ W/(Km)

summary

computer simulation of SiO₂ melts and glasses:

- low temperatures: c_V in harmonic approximation using $g(\omega)$ from classical MD simulation
- frequency-dependent specific heat allows to determine the contribution of the configurational degrees of freedom to the static specific heat
- from $\Phi_{TT}(q,t)$ one can extract the thermal conductivity

G. S. Grest, S. R. Nagel, *J. Phys. Chem.* **91**, 4916 (1987);

W. Götze, A. Latz, *J. Phys.: Condens. Matter* **1**, 4169 (1989);

J. K. Nielsen, *Phys. Rev. E* **60**, 471 (1999);

J. Horbach, W. Kob, K. Binder, *J. Phys. Chem. B* **103**, 4104 (1999);

P. Scheidler, W. Kob, A. Latz, J. Horbach, K. Binder, *Phys. Rev. B* **63**, 104204 (2001).

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